

### 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

### SECTION 1 Identification of the substance / mixture and of the company / undertaking

Product Name 5-BROMONICOTINIC ACID

Chemical Name Not Available

Synonyms C6-H4-Br-N-O2; nicotinic acid, 5-bromo-; 5-bromo-3-pyridinecarboxylic acid

Chemical Formula C6-H4-Br-N-O2
Other means of identification
CAS Number 20826-04-4

#### Relevant identified uses of the substance or mixture and uses advised against:

Relevant identified uses	Intermediate
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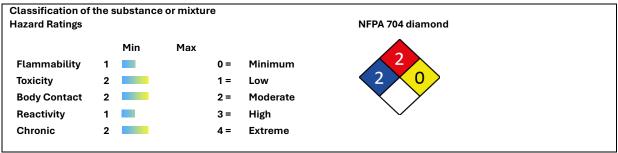
## $\label{lem:decomposition} \textbf{Details of the manufacturer or supplier of the safety data sheet:}$

Registered company name	Cohizon Life Sciences Limited
Address	Plot No. 6102/3, 6117-19, 5809-10, GIDC, Ankleshwar Gujarat 393002 India
Telephone	Not Available
Fax	Not Available
Website	Not Available
Email	Not Available

#### Emergency telephone number:

Association / Organisation	Cohizon Life Sciences Limited
Emergency telephone number(s)	(+91) 7046611150/51
Other emergency telephone number(s)	(+91) 6357684904

### **SECTION 2 Hazards identification**



Note: The hazard category numbers found in GHS classification in section 2 of this SDSs are NOT to be used to fill in the NFPA 704 diamond. Blue = Health Red = Fire Yellow = Reactivity White = Special (Oxidizer or water reactive substances).

	Acute Toxicity (Oral) Category 4, Skin Corrosion/Irritation Category 2, Sensitisation
	(Skin) Category 1, Serious Eye Damage/Eye Irritation Category 2A, Specific Target
Classification	Organ Toxicity - Single Exposure (Respiratory Tract Irritation) Category 3, Specific Target
	Organ Toxicity - Repeated Exposure Category 2, Hazardous to the Aquatic Environment
	Long-Term Hazard Category 3

### Label elements

Hazard pictogram(s)	<b>!</b>	
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### 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

Signal word	Warning
Hazard statement(s)	
H302	Harmful if swallowed.
H315	Causes skin irritation.
H317	May cause an allergic skin reaction.
H319	Causes serious eye irritation.
H335	May cause respiratory irritation.
H373	May cause damage to organs through prolonged or repeated exposure.
H/112	Harmful to aquatic life with long lasting effects

### Precautionary statement(s) Prevention

,	
P260	Do not breathe dust/fume.
P271	Use only outdoors or in a well-ventilated area.
P280	Wear protective gloves, protective clothing, eye protection and face protection.
P264	Wash all exposed external body areas thoroughly after handling.
P270	Do not eat, drink or smoke when using this product.
P273	Avoid release to the environment.
P272	Contaminated work clothing should not be allowed out of the workplace.

#### Precautionary statement(s) Response

Treductionary statement(s) nesponse	
P302+P352	IF ON SKIN: Wash with plenty of water.
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if
	present and easy to do. Continue rinsing.
P333+P313	If skin irritation or rash occurs: Get medical advice/attention.
P337+P313	If eye irritation persists: Get medical advice/attention.
P362+P364	Take off contaminated clothing and wash it before reuse.
P301+P312	IF SWALLOWED: Call a POISON CENTER/doctor/physician/first aider if you feel unwell.
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.
P330	Rinse mouth.

### Precautionary statement(s) Storage

P403+P235	Store in a well-ventilated place. Keep cool.
P405	Store locked up.

### Precautionary statement(s) Disposal

P501	Dispose of contents/container to authorised hazardous or special waste collection point
	in accordance with any local regulation

## SECTION 3 Composition / information on ingredients

## Substances

Cas No.	%[weight]	Name
20826-04-4	>98	5-bromonicotinic acid

### Mixtures

See Section above for composition of substances

# SECTION 4 First aid measures

# Description of first aid measures

	If this product comes in contact with the eyes:
Eye Contact	Wash out immediately with fresh running water.
	<ul> <li>Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.</li> </ul>
	<ul> <li>Seek medical attention without delay; if pain persists or recurs seek medical attention.</li> </ul>



#### 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

	<ul> <li>Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.</li> </ul>
	If skin or hair contact occurs:
01: 0	<ul> <li>Immediately remove all contaminated clothing, including footwear.</li> </ul>
Skin Contact	Flush skin and hair with running water (and soap if available
	Seek medical attention in event of irritation
	<ul> <li>If fumes or combustion products are inhaled remove from contaminated area.</li> <li>Lay patient down. Keep warm and rested.</li> </ul>
	Prostheses such as false teeth, which may block airway, should be removed,
	where possible, prior to initiating first aid procedures.
Inhalation	Apply artificial respiration if not breathing, preferably with a demand valve
	resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.
	Transport to hospital, or doctor, without delay.
	IF SWALLOWED, REFER FOR MEDICAL ATTENTION, WHERE POSSIBLE, WITHOUT DELAY.
	<ul> <li>For advice, contact a Poisons Information Centre or a doctor.</li> </ul>
	<ul> <li>Urgent hospital treatment is likely to be needed</li> </ul>
	<ul> <li>In the mean time, qualified first-aid personnel should treat the patient</li> </ul>
	following observation and employing supportive measures as indicated by the patient's condition.
	If the services of a medical officer or medical doctor are readily available, the
	patient should be placed in his/her care and a copy of the SDS should be
Ingestion	provided. Further action will be the responsibility of the medical specialist.
	If medical attention is not available on the worksite or surroundings send the      action to a least it to set to available on the worksite or surroundings send the      action to a least it to set to available on the worksite or surroundings send the
	patient to a hospital together with a copy of the SDS.
	Where medical attention is not immediately available or where the
	patient is more than 15 minutes from a hospital or unless instructed
	otherwise:
	INDUCE vomiting with fingers down the back of the throat, ONLY IF  CONSCIOUS: Learn action forward or place any left side. (head down.)
	CONSCIOUS. Lean patient forward or place on left side (head-down
	position, if possible) to maintain open airway and prevent aspiration.
	NOTE: Wear a protective glove when inducing vomiting by mechanical means

### Indication of any immediate medical attention and special treatment needed

As in all cases of suspected poisoning, follow the ABCDEs of emergency medicine (airway, breathing, circulation, disability, exposure), then the ABCDEs of toxicology (antidotes, basics, change absorption, change distribution, change elimination).

For poisons (where specific treatment regime is absent):

### **Basic Treatment:**

- Establish a patent airway with suction where necessary.
- Watch for signs of respiratory insufficiency and assist ventilation as necessary
- Administer oxygen by non-rebreather mask at 10 to 15 L/min
- Monitor and treat, where necessary, for pulmonary oedema
- Monitor and treat, where necessary, for shock
- Anticipate seizures
- **DO NOT** use emetics. Where ingestion is suspected rinse mouth and give up to 200 ml water (5 ml/kg recommended) for dilution where patient is able to swallow, has a strong gag reflex and does not drool.

#### **Advanced Treatment:**

- Consider orotracheal or nasotracheal intubation for airway control in unconscious patient or where respiratory arrest has occurred.
   Positivepressure ventilation using a bag-valve mask might be of use.
- Monitor and treat, where necessary, for arrhythmias.
- Start an IV D5W TKO. If signs of hypovolaemia are present use lactated Ringers solution. Fluid overload might create complications. Drug therapy should be



#### 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

considered for pulmonary oedema.

- Hypotension with signs of hypovolaemia requires the cautious administration
  of fluids. Fluid overload might create complications. Treat seizures with
  diazepam.
- Proparacaine hydrochloride should be used to assist eye irrigation.

BRONSTEIN, A.C. and CURRANCE, P.L.

EMERGENCY CARE FOR HAZARDOUS MATERIALS EXPOSURE: 2nd Ed. 1994

Treat symptomatically

### **SECTION 5 Firefighting measures**

### **Extinguishing media**

- Foam
- Dry Chemical Powder
- BCF (where regulations permit).
- Carbon dioxide

Fire/Explosion Hazard

Water Spray or Fog – Larges files only

Special hazards arising from the substrate or mixture

Fire Incompatibility	Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches,
	pool chlorine etc. as ignition may result.

# Advice for firefighters Alert Fire Brigade and tell them location and nature of hazard. Wear breathing apparatus plus protective gloves Prevent, by any means available, spillage from entering drains or water courses Fight fire from a safe distance, with adequate cover. Fire Fighting Use water delivered as a fine spray to control fire and cool adjacent area DO NOT approach containers suspected to be hot Cool fire exposed containers with water spray from a protected location If safe to do so, remove containers from path of fire. Equipment should be thoroughly decontaminated after use. Combustible solid which burns but propagates flame with difficulty; it is estimated that most organic dusts are combustible (circa 70%) - according to the circumstances under which the combustion process occurs, such materials may cause fires and / or dust explosions. Organic powders when finely divided over a range of concentrations

MEC).

# regardless of particulate size or shape and suspended in air or some other oxidizing medium may form explosive dust-air mixtures and result in a fire or dust explosion (including secondary explosions). Avoid generating dust, particularly clouds of dust in a confined or unventilated space as dusts may form an explosive mixture with air, and any source of ignition, i.e. flame or spark, will cause fire or explosion. Dust clouds generated by the fine grinding of the solid are a particular hazard; accumulations of fine dust (420 micron or less) may burn rapidly and fiercely if ignited - particles exceeding this limit will generally not form flammable dust clouds; once initiated, however, larger particles up to 1400 microns diameter will contribute to the propagation of an explosion. In the same way as gases and vapours, dusts in the form of a cloud are only ignitable over a range of concentrations; in principle, the concepts of lower explosive limit (LEL) and upper explosive limit (UEL) are applicable to dust clouds but only the LEL is of practical use; - this is because of the inherent difficulty of achieving homogeneous dust clouds at high temperatures (for dusts the LEL is often called the "Minimum Explosible Concentration",



#### 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

- When processed with flammable liquids/vapors/mists,ignitable (hybrid) mixtures may be formed with combustible dusts. Ignitable mixtures will increase the rate of explosion pressure rise and the Minimum Ignition Energy (the minimum amount of energy required to ignite dust clouds MIE) will be lower than the pure dust in air mixture. The Lower Explosive Limit (LEL) of the vapour/dust mixture will be lower than the individual LELs for the vapors/mists or dusts.
- A dust explosion may release of large quantities of gaseous products; this in turn creates a subsequent pressure rise of explosive force capable of damaging plant and buildings and injuring people.
- Usually the initial or primary explosion takes place in a confined space such
  as plant or machinery, and can be of sufficient force to damage or rupture
  the plant. If the shock wave from the primary explosion enters the
  surrounding area, it will disturb any settled dust layers, forming a second
  dust cloud, and often initiate a much larger secondary explosion. All large
  scale explosions have resulted from chain reactions of this type.
- Dry dust can be charged electrostatically by turbulence, pneumatic transport, pouring, in exhaust ducts and during transport.
- build-up of electrostatic charge may be prevented by bonding and grounding.
- Powder handling equipment such as dust collectors, dryers and mills may require additional protection measures such as explosion venting.
- All movable parts coming in contact with this material should have a speed of less than 1-meter/sec.
- A sudden release of statically charged materials from storage or process equipment, particularly at elevated temperatures and/ or pressure, may result in ignition especially in the absence of an apparent ignition source.
- One important effect of the particulate nature of powders is that the surface
  area and surface structure (and often moisture content) can vary widely
  from sample to sample, depending of how the powder was manufactured
  and handled; this means that it is virtually impossible to use flammability
  data published in the literature for dusts (in contrast to that published for
  gases and vapours).
- Autoignition temperatures are often quoted for dust clouds (minimum ignition temperature (MIT)) and dust layers (layer ignition temperature (LIT)); LIT generally falls as the thickness of the layer increases
- Combustion products include.
- carbon monoxide (CO)
- carbon dioxide (CO2)
- Hydrogen bromide
- nitrogen oxides (NOx)
- other pyrolysis products typical of burning organic material.
- May emit poisonous fumes.
- May emit corrosive fumes.

### **SECTION 6 Accidental release measures**

Personal precautions, protective equipment and emergency procedures

See section 8

**Environmental precautions** 

See section 12

### Methods and material for containment and cleaning up

	•	Remove all ignition sources.
Minor Spills	•	Clean up all spills immediately.
	•	Avoid contact with skin and eyes.



## 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

	Control personal contact with the substance, by using protective equipment.
	Use dry clean up procedures and avoid generating dust.
	Place in a suitable, labelled container for waste disposal.
	Moderate hazard.
	CAUTION: Advise personnel in area.
	Alert Emergency Services and tell them the location and nature of hazard.
	<ul> <li>Control personal contact by wearing protective clothing.</li> </ul>
	<ul> <li>Prevent, by any means available, spillage from entering drains or water courses.</li> </ul>
Majar Spilla	Recover products wherever possible.
Major Spills	IF DRY: Use dry clean up procedures and avoid generating dust. Collect residues and place in sealed plastic bags or other containers for disposal.
	IF WET: Vacuum/shovel up and place in labelled containers for disposal.
	ALWAYS: Wash area down with large amounts of water and prevent runoff into drains.
	If contamination of drains or waterways occurs, advise Emergency Services.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

## SECTION 7 Handling and storage

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Precautions for safe handling			
	Avoid all personal contact, including inhalation.		
	Wear protective clothing when risk of exposure occurs.		
	Use in a well-		
	ventilated area.		
	Prevent concentration in hollows and sumps.		
	DO NOT enter confined spaces until the atmosphere has been checked.		
	DO NOT allow material to contact humans, exposed food or food utensils		
	Avoid contact with incompatible		
	materials.		
	When handling, <b>DO NOT</b> eat, drink or smoke.		
	Always wash hands with soap and water after handling.		
	Work clothes should be laundered separately.		
	Launder contaminated clothing before re-use.		
	Use good occupational work practice.		
	Observe manufacturer's storage and handling recommendations contained		
	within this SDS.		
	Atmosphere should be regularly checked against established exposure		
Safe Handling	standards to ensure safe working conditions are maintained.		
	Organic powders when finely divided over a range of concentrations		
	regardless of particulate size or shape and suspended in air or some other		
	oxidizing medium may form explosive dust-air mixtures and result in a fire or		
	dust explosion (including secondary explosions)		
	Minimise airborne dust and eliminate all ignition sources. Keep away from		
	heat, hot surfaces, sparks, and flame		
	Establish good housekeeping practices.		
	Remove dust accumulations on a regular basis by vacuuming or gentle		
	sweeping to avoid creating dust clouds.		
	Use continuous suction at points of dust generation to capture and minimise		
	the accumulation of dusts. Particular attention should be given to overhead		
	and hidden horizontal surfaces to minimise the probability of a "secondary"		
	explosion. According to NFPA Standard 654, dust layers 1/32 in.(0.8 mm)		
	thick can be sufficient to warrant immediate cleaning of the area.		
	Do not use air hoses for cleaning.		
	Minimise dry sweeping to avoid generation of dust clouds. Vacuum dust-		

with explosion-proof motors should be used.

accumulating surfaces and remove to a chemical disposal area. Vacuums



## 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

	<ul> <li>Control sources of static electricity. Dusts or their packages may accumulate static charges, and static discharge can be a source of ignition.</li> <li>Solids handling systems must be designed in accordance with applicable standards (e.g. NFPA including 654 and 77) and other national guidance.</li> <li>Do not empty directly into flammable solvents or in the presence of flammable vapors.</li> <li>The operator, the packaging container and all equipment must be grounded with electrical bonding and grounding systems. Plastic bags and plastics cannot be grounded, and antistatic bags do not completely protect against development of static charges.</li> <li>Empty containers may contain residual dust which has the potential to accumulate following settling. Such dusts may explode in the presence of an appropriate ignition source.</li> <li>Do NOT cut, drill, grind or weld such containers.</li> <li>In addition ensure such activity is not performed near full, partially empty or empty containers without appropriate workplace safety authorisation or permit.</li> </ul>
Other information	<ul> <li>Store in original containers.</li> <li>Keep containers securely sealed.</li> <li>Store in a cool, dry area protected from environmental extremes.</li> <li>Store away from incompatible materials and foodstuff containers.</li> <li>Protect containers against physical damage and check regularly for leaks.</li> <li>Observe manufacturer's storage and handling recommendations contained within this SDS</li> <li>For major quantities:</li> <li>Consider storage in bunded areas - ensure storage areas are isolated from sources of community water (including stormwater, ground water, lakes and streams).</li> <li>Ensure that accidental discharge to air or water is the subject of a contingency disaster management plan; this may require</li> <li>consultation with local authorities.</li> </ul>

### Conditions for safe storage, including any incompatibilities

Suitable container	<ul> <li>Polyethylene or polypropylene container.</li> <li>Check all containers are clearly labelled and free from leaks.</li> </ul>	
Storage incompatibility	Avoid reactions with oxidising agents.	

# **SECTION 8 Exposure controls / personal protection**

### **Control Parameters**

Occupational Exposure Limits (OEL)

## INGREDIENT DATA

Not Available

Ingredient	Original IDLH	Revised IDLH
5-bromonicotinic acid	Not Available	Not Available

### **Occupational Exposure Banding**

occupational Exposure Burlaing			
Ingredient	Occupational Exposure Band Rating	Occupational Exposure Band Limit	
5-bromonicotinic acid	D	> 0.01 to ≤ 0.1 mg/m <sup>3</sup>	
Notes:		potency and the adverse health outcomes is process is an occupational exposure band	

### **Exposure controls**

Appropriate engineering	•	Engineering controls are used to remove a hazard or place a barrier
controls		between the worker and the hazard. Well-designed engineering



# 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

	controls can be highly effective in pr	otecting workers	and will typically	
	be independent of worker interactions to provide this high level of			
	protection.			
	<ul> <li>The basic types of engineering contr</li> </ul>			
	Process controls which involve char	nging the way a jo	b activity or process	
	is done to reduce the risk.			
	Enclosure and/or isolation of emissi		•	
	hazard "physically" away from the w			
	strategically "adds" and "removes" a			
	Ventilation can remove or dilute an a		=	
	properly. The design of a ventilation process and chemical or contamina	-	•	
	use multiple types of controls to pre	-	-	
	For flammable liquids and flammable	· ·	· · · · · · · · · · · · · · · · · · ·	
	process enclosure ventilation syster	_		
	equipment should be explosion-resi		a. Vontitution	
	Air contaminants generated in the w		s varving "escape"	
	velocities which, in turn, determine t			
	circulating air required to effectively	•		
	J 1,1 11 11 11 11 11 11 11 11 11 11 11 11			
	Type of Contaminant		Air Speed	
	solvent, vapours, degreasing etc., evaporating	ng from tank (in	1-2.5 m/s (200-	
	still air).		500 ft/min)	
	aerosols, fumes from pouring operation		2.5-10 m/s (500-	
	container filling, low speed conveyer transfers		2000 f/min.)	
	drift, plating acid fumes, pickling (released at l zone of active generation)	ow velocity into		
	Which each range the appropriate value dep	ends on		
	Lower end of the range Upper end of the range			
	1. Room air currents minimal or favourable 1. Disturbing room air currents			
	to capture  2. Contaminants of law toxisity or of 2. Contaminants of high toxisity			
	2. Contaminants of low toxicity or of nuisance value only			
	3. Intermittent, low production 3: High production, heavy use			
	4. Large hood or large air mass in motion 4: Small hood-local control only			
	Simple theory shows that air velocity falls rapidly with distance away from			
	the opening of a simple extraction pi	pe. Velocity gene	erally decreases with	
	the square of distance from the extra	action point (in si	mple cases).	
	Therefore, the air speed at the extra	•		
	accordingly, after reference to distar		=	
	The air velocity at the extraction fan			
	of 4-10 m/s (800-2000 ft/min) for ext		_	
	metres distant from the extraction perconsiderations, producing performa			
	apparatus, make it essential that the			
	factors of 10 or more when extraction			
Individual protection		.,		
measures, such as personal				
protective				
equipment				
	Safety glasses with side shields.			
	Chemical goggles. [AS/NZS 1337.1, EN166 or national equivalent]			
	Contact lenses may pose a special hazard; soft contact lenses may			
	absorb and concentrate irritants. A written policy document, describing			
Eye and face protection	the wearing of lenses or restrictions	on use, should b	e created for each	
	workplace or task. This should inclu	ıde a review of ler	ns absorption and	
	adsorption for the class of chemicals in use and an account of injury			
	experience.			



# 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

<u></u>	Madical and fine address and a band discharge of the shall not be a		
	Medical and first-aid personnel should be trained in their removal and  suitable agruinment should be readily available. In the great of		
	suitable equipment should be readily available. In the event of		
	chemical exposure, begin eye irrigation immediately and remove		
	contact lens as soon as practicable. Lens should be removed at the		
	first signs of eye redness or irritation - lens should be removed in a		
	clean environment only after workers have washed hands thoroughly.		
	[CDC NIOSH Current Intelligence Bulletin 59].		
Skin protection	See Hand protection below		
	NOTE:		
	<ul> <li>The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other</li> </ul>		
	<ul> <li>protective equipment, to avoid all possible skin contact.</li> <li>Contaminated leather items, such as shoes, belts and watch-bands</li> </ul>		
	should be removed and destroyed		
	The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer.  Where the chemical is a preparation of several substances, the resistance of		
	the glove material can not be calculated in advance and has therefore to be checked prior to the application.		
	The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final		
	choice.		
	Personal hygiene is a key element of effective hand care. Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly.		
	Application of a non-perfumed moisturizer is recommended.  Suitability and durability of glove type is dependent on usage. Important factors in		
	the selection of gloves include:		
	frequency and duration of contact,		
	chemical resistance of glove material,		
	glove thickness and		
Hands/feet protection	• dexterity		
	Select gloves tested to a relevant standard (e.g. Europe EN 374, US F739, AS/NZS 2161.1 or national equivalent).		
	<ul> <li>When prolonged or frequently repeated contact may occur, a glove with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to EN 374, AS/NZS 2161.10.1 or</li> </ul>		
	national equivalent) is recommended.		
	When only brief contact is expected, a glove with a protection class		
	of 3 or higher (breakthrough time greater than 60 minutes according to EN 374, AS/NZS 2161.10.1 or national equivalent) is		
	recommended.		
	Some glove polymer types are less affected by movement and this should be taken into account when considering gloves for long-term		
	use.  • Contaminated gloves should be replaced.		
	As defined in ASTM F-739-96 in any application, gloves are rated as:		
	Excellent when breakthrough time > 480 min		
	Good when breakthrough time > 20 min		
	Fair when breakthrough time < 20 min		
	Poor when glove material degrades		
	For general applications, gloves with thickness typically greater than 0.35 mm, are recommended.		
	It should be emphasized that glove thickness is not necessarily a good predictor of		



### 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

	glove resistance to a specific chemical, as the permeation efficiency of the glove			
	will be dependent on the exact composition of the glove material. Therefore, glove			
	selection should also be based on consideration of the task requirements and			
	knowledge of breakthrough times.			
	Glove thickness may also vary depending on the glove manufacturer, the glove type			
	and the glove model. Therefore, the manufacturers' technical data should always be			
	taken into account to ensure selection of the most appropriate glove for the task.			
	Note: Depending on the activity being conducted, gloves of varying thickness may			
	be required for specific tasks. For example:			
	<ul> <li>Thinner gloves (down to 0.1 mm or less) may be required where a high</li> </ul>			
	degree of manual dexterity is needed. However, these gloves are only			
	likely to give short duration protection and would normally be just for			
	single use applications, then disposed of.			
	<ul> <li>Thicker gloves (up to 3 mm or more) may be required where there is a</li> </ul>			
	mechanical (as well as a chemical) risk i.e. where there is abrasion or			
	puncture potential			
	Gloves must only be worn on clean hands. After using gloves, hands should be			
	washed and dried thoroughly. Application of a non-perfumed moisturiser is			
	recommended.			
	Experience indicates that the following polymers are suitable as glove materials			
	for protection against undissolved, dry solids, where abrasive particles are not			
	present.			
	<ul> <li>polychloroprene.</li> </ul>			
	nitrile rubber.			
	butyl rubber			
	fluor caoutchouc			
	polyvinyl chloride			
	Gloves should be examined for wear and/ or degradation constantly.			
Body protection	See Other protection below			
	Overalls.			
	P.V.C apron			
Other protection	Barrier cream.			
	Skin cleansing cream.			
	Eye wash unit			
	L			

### **Respiratory protection**

Type -P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent).

Required Minimum	Half-Face Respirator	Full-Face Respirator	Powered Air Respirator
Protection Factor			
up to 10 x ES	P1	-	PAPR-P1
	Air-line*	-	-
up to 50 x ES	Air-line**	P2	PAPR-P2
up to 100 x ES	-	P3	-
		Air-line*	-
100+ x ES	-	Air-line**	PAPR-P3

<sup>\*</sup>Negative pressure demand \*\* - Continuous flow

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

- Respirators may be necessary when engineering and administrative controls do not adequately prevent exposures.
- The decision to use respiratory protection should be based on professional judgment that takes into account toxicity information, exposure measurement data, and frequency and likelihood of the worker's exposure ensure users are not subject to high thermal loads which may result in heat stress



### 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

- or distress due to personal protective equipment (powered, positive flow, full face apparatus may be
- Published occupational exposure limits, where they exist, will assist in determining the adequacy of the selected respiratory protection. These may be government mandated or vendor recommended.
- Certified respirators will be useful for protecting workers from inhalation of particulates when properly selected and fit tested as part of a complete respiratory protection program.
- Where protection from nuisance levels of dusts are desired, use type N95 (US) or type P1 (EN143) dust masks. Use respirators and components tested and approved under appropriate government standards such as NIOSH (US) or CEN (EU)
- Use approved positive flow mask if significant quantities of dust becomes airborne.
- Try to avoid creating dust conditions.

### SECTION 9 Physical and chemical properties

Appearance	Tan powder; does not i	mix well with water	
Physical state	Divided Solid	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n- octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Applicable	Decomposition temperature (°C)	Not Available
Melting point / freezing point (°C)	178-180°C	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	202.1
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	<b>Explosive properties</b>	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Negligible
Vapour pressure (kPa)	Negligible	Gas group	Not Available
Solubility in water	Partially miscible	pH as a solution (1%)	Not Applicable
Vapour density (Air = 1)	Not Available	VOC g/L	Not Applicable
Heat of Combustion (kJ/g)	Not Available	Ignition Distance (cm)	Not Available
Flame Height (cm)	Not Available	Flame Duration (s)	Not Available
Enclosed Space Ignition Time		Enclosed Space	
Equivalent (s/m3)	Not Available	Ignition Deflagration	Not Available
		Density (g/m3)	

# **SECTION 10 Stability and reactivity**

Reactivity	See section 7	
Chemical stability	Unstable in the presence of incompatible materials	
	<ul> <li>Product is considered stable</li> </ul>	
	Hazardous polymerisation will not occur	
Possibility of hazardous reactions	See section 7	
Conditions to avoid	See section 7	
Incompatible materials	See section 7	
Hazardous decomposition products	See section 5	



# 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

# SECTION 11 Toxicological information

### Information on toxicological effects

Information on toxicologi	ical effects
	The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage.
	Persons with impaired respiratory function, airway diseases and conditions such as emphysema or chronic bronchitis, may incur further disability if excessive concentrations of particulate are inhaled.
Inhaled	If prior damage to the circulatory or nervous systems has occurred or if kidney damage has been sustained, proper screenings should be conducted on individuals who may be exposed to further
	risk if handling and use of the material result in excessive exposures.
	Pyridine and its derivatives generally produce local irritation on contact with the mucous
	membranes. Overexposure to pyridine  • and some of its derivatives may produce headache, nausea, loss of consciousness,
	nervousness, loss of appetite, sleeplessness and narcosis
	Accidental ingestion of the material may be harmful; animal experiments
	indicate that ingestion of less than 150 gram may be fatal or may produce
	serious damage to the health of the individual.
Ingestion	Nicotinic acid (vitamin B3) and some of its derivatives may dilate the
	blood vessels. When given by mouth or by injection during treatment, it
	may produce a temporary flushing of the face, a sensation of heat and a
	pounding in the head.      This material can cause inflammation of the skin on contact in some
	persons.
	The material may accentuate any pre-existing dermatitis condition
	Skin contact with the material may damage the health of the individual;
	systemic effects may result following absorption.
	<ul> <li>Open cuts, abraded or irritated skin should not be exposed to this material.</li> </ul>
Skin Contact	<ul> <li>Entry into the blood-stream, through, for example, cuts, abrasions or</li> </ul>
	lesions, may produce systemic injury with harmful effects.
	Examine the skin prior to the use of the material and ensure that any
	external damage is suitably protected.
	<ul> <li>Pyridine and derivatives cause local irritation to the skin; absorption</li> </ul>
	through the skin can cause similar effects as inhalation.
Eye	<ul> <li>This material can cause eye irritation and damage in some persons.</li> <li>Pyridine and its derivatives generally produce local irritation on contact with the cornea.</li> </ul>
	Repeated or long-term occupational exposure is likely to produce cumulative
	health effects involving organs or biochemical systems.
	Long-term exposure to respiratory irritants may result in airways disease, involving
	difficulty breathing and related whole-body problems.
	Skin contact with the material is more likely to cause a sensitisation reaction in
	some persons compared to the general population.
	There has been some concern that this material can cause cancer or mutations
	but there is not enough data to make an assessment.
	There is some evidence from animal testing that exposure to this material may
Chronic	result in toxic effects to the unborn baby.  Data from experimental studies indicate that pyridines represent a potential
Chronic	cause of cancer in man. They have also been shown to cross the placental barrier
	in rats and cause premature delivery, miscarriages and stillbirths.
	Long term exposure to high dust concentrations may cause changes in lung
	function i.e. pneumoconiosis, caused by particles less than 0.5 micron
	penetrating and remaining in the lung.
	Chronic poisoning from ionic bromides has historically resulted from medical use
	of bromides but not from exposure in the environment or workplace. In the
	absence of other signs of poisoning, there may be depression, hallucinations and
	schizophrenia-like psychosis. Bromides may also cause sedation, irritability,
	agitation, delirium, memory loss, confusion, disorientation, forgetfulness,



#### 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

inability to speak, difficulty speaking, weakness, fatigue, a spinning sensation,
stupor, coma, decreased appetite, nausea, vomiting, an acne-like rash on the
face (bronchoderma), legs and trunk, swelling of the bronchi and a profuse
discharge from the nostrils. There may also be inco-ordination and very brisk
reflexes. Correlation of nervous system symptoms with blood levels of bromide is
inexact. Current day usage of bromides is generally limited to antihistamines
such as brompheniramine, which is a covalent compound; ionic compounds are
no longer regularly used due to their toxicity.
In test animals, brominated vegetable oils (BVOs), historically used as emulsifiers in
certain soda-based soft drinks, produced damage to the heart and kidneys in
addition to increasing fat deposits in these organs. In extreme cases, BVOs caused
testicular damage, stunted growth and produced lethargy and fatigue.
Brominism (chronic bromine poisoning) produces slurred speech, apathy,
headache, decreased memory, anorexia and drowsiness, psychosis resembling
paranoid schizophrenia, and personality changes.
Several cases of foetal abnormalities have been described in mothers who took
large doses of bromides during pregnancy. Reproductive effects caused by bromide
(which crosses the placenta) include central nervous system depression,
brominism, and bronchoderma (an acne-like rash) in the newborn.

5-BROMONICOTINIC ACID	TOXICITY	IRRITATION		
	Not Available	Not Available		
Legend:	1. Value obtained from Eur	1. Value obtained from Europe ECHA Registered Substances - Acute toxicity		
	2. Value obtained from ma	2. Value obtained from manufacturer's SDS. Unless otherwise specified data		
	extracted from RTECS - Rea	extracted from RTECS - Register of Toxic Effect of chemical Substances		

Ptosis, somnolence recorded.

The following information refers to contact allergens as a group and may not be specific to this product.

Contact allergies quickly manifest themselves as contact eczema, more rarely as urticaria or Quincke's oedema. The pathogenesis of contact eczema involves a cell-mediated (T lymphocytes) immune reaction of the delayed type. Other allergic skin reactions, e.g. contact urticaria, involve antibody-mediated immune reactions. The significance of the contact allergen is not simply determined by its sensitization potential: the distribution of the substance and the opportunities for contact with it are equally important. A weakly sensitizing substance which is widely distributed can be a more important allergen than one with stronger sensitizing potential with which few individuals come into contact. From a clinical point of view, substances are noteworthy if they produce an allergic test reaction in more than 1% of the persons tested.

5-BROMONICOTINIC ACID

Asthma-like symptoms may continue for months or even years after exposure to the material ends. This may be due to a non- allergic condition known as reactive airways dysfunction syndrome (RADS) which can occur after exposure to high levels of highly irritating compound. Main criteria for diagnosing RADS include the absence of previous airways disease in a non-atopic individual, with sudden onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. Other criteria for diagnosis of RADS include a reversible airflow pattern on lung function tests, moderate to severe bronchial hyperreactivity on methacholine challenge testing, and the lack of minimal lymphocytic inflammation, without eosinophilia. RADS (or asthma) following an irritating inhalation is an infrequent disorder with rates related to the concentration of and duration of exposure to the irritating substance. On the other hand, industrial bronchitis is a disorder that occurs as a result of exposure due to high concentrations of irritating substance (often particles) and is completely reversible after exposure ceases. The disorder is characterized by difficulty breathing, cough and mucus production.



#### 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

Acute Toxicity	~	Carcinogenicity	×	
Skin Irritation/Corrosion	•	Reproductivity	×	
Serious Eye Damage/Irritation	~	STOT - Single Exposure	<b>~</b>	
Respiratory or Skin sensitisation	•	STOT - Repeated Exposure	•	
Mutagenicity	×	Aspiration Hazard	×	
Legand: X – Data either not availa		r does not fill the criteria for classification		

### **SECTION 12 Ecological information**

#### Toxicity

	End Point	Test Duration	Species	Value	Source
5-BROMONICOTINIC ACID	Not	(hr.) Not Available	Not Available	Not	Not
	Available	110t7tVallable	14017 Wallable	Available	Available
Legand:	Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity				
	4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data				

Harmful to aquatic organisms.

May cause long-term adverse effects in the aquatic environment.

Do NOT allow product to come in contact with surface waters or to intertidal areas below the mean high water mark. Do not contaminate water when cleaning equipment or disposing of equipment wash-waters.

Wastes resulting from use of the product must be disposed

of on site or at approved waste sites. For Bromide:

Environmental Fate: Bromide ions may be introduced to the environment after the breakdown of various salts and complexes or after the degradation of organic compounds that contain carbon bonded to bromine. Bromides may also affect the growth of micro-organisms and have been used for this purpose in industry. Bromides in drinking water are occasionally subject to disinfection processes involving ozone of chlorine. Bromide may be oxidize to produce hypobromous acid which in turn may react with natural organic matter to form brominated compounds. Bromates may also be formed following ozonation or chlorination if pH is relatively high.

Atmospheric Fate: Hydrogen bromide (HBr) and bromine nitrate (BrONO2), are much more easily broken up by sunlight causing bromine to be from 10 to 100 times more effective than chlorine at destroying ozone. From 30-60% of bromocarbons released to the atmosphere are man-made (methyl bromide fumigants and halon fire extinguishers) and both compounds are restricted by international agreement.

Ecotoxicity: Bromates may be animal carcinogens. Although not a significant toxin in mammalian or avian systems it is highly toxic to rainbow trout and Daphnia magna. On the average, sodium bromide is highly toxic to blugill, rainbow trout, sheepdhead minnow, water fleas and mysid shrimp. Bromides have a negative effect on the growth and development of oyster species.

For Pyridine and its Derivatives:

Environmental Fate: As molecular weight/substitution increase in the pyridine category, greater distribution to water and soil, and less to air, is predicted. Atmospheric Fate: The lower weight pyridine, piperidine, is expected to be rapidly degraded by UV light in the atmosphere, with an estimated half-life of < 1 day. Higher molecular weight pyridines are expected to be broken down by sunlight, (photodegrades), more slowly, (half-lives ranging from 10-30 days). Lutidines and collidines are expected to photodegrade even more slowly. The nitrile derivatives of pyridine are also predicted to photodegrade slowly, with half-lives of 164 days; however, the nitrile derivatives of pyridine are predicted to partition to air much less favorably than to soil and water.

Terrestrial Fate: Depending upon the environmental conditions, different types of bacteria, fungi, and enzymes are involved in the breakdown of these substances. Aquatic Fate: Pyridines are not expected to be broken down by water; however, breakdown by microbes is expected, if sufficient oxygen is available. These substances are expected to be stable in low oxygen/sterile conditions.

Ecotoxicity: Pyridine and its derivatives range from slightly to moderately toxic to fish, invertebrates and algae.



#### 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

### DO NOT discharge into sewer or waterways

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
5-BROMONICOTINIC ACID	High	High

#### Bio accumulative potential

Ingre	dient	Bioaccumulation
5-BR0	OMONICOTINIC ACID	Low (BCF = 1.58)

#### Mobility in soil

Ingredient	Mobility
5-BROMONICOTINIC ACID	LOW (Log KOC = 23.47)

### **SECTION 13 Disposal considerations**

#### Waste treatment methods

- Containers may still present a chemical hazard/ danger when empty
- Return to supplier for reuse/ recycling if possible.

#### Otherwise:

- If container cannot be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised lendfill.
- Where possible retain label warnings and SDS and observe all notices pertaining to the product

Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.

A Hierarchy of Controls seems to be common - the user should investigate:

- Reduction
- Reuse
- Recycling
- Disposal (if all else fails)

# Product / Packaging disposal

This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate. In most instances the supplier of the material should be consulted.

- DO NOT allow wash water from cleaning or process equipment to enter drains
- It may be necessary to collect all wash water for treatment before disposal
- In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.
- Where in doubt contact the responsible authority
- Recycle wherever possible.
- Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.
- Dispose of by: burial in a land-fill specifically licensed to accept chemical and / or pharmaceutical wastes or Incineration in a licensed apparatus (after admixture with suitable combustible material).
- Decontaminate empty containers. Observe all label safeguards until containers are cleaned and destroyed

# **SECTION 14 Transport information**

#### **Labels Required**

Marine Pollutant	No



5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

Land transport (UN): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Air transport (ICAO-IATA / DGR): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

Sea transport (IMDG-Code/ GGVSee): NOT REGULATED FOR TRANSPORT OF DANGEROUS GOODS

#### 14.7.1 Transport in bulk in accordance to Annex II of MARPOL and the IBC Code

Not Applicable

14.7.2 Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product Name	Ship Type
5-BROMONICOTINIC ACID	Not Available

#### 14.7.3 Transport in bulk in accordance with the IGC Code

Product Name	Ship Type	
5-BROMONICOTINIC ACID	Not Available	

## SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

### 5-BROMONICOTINIC ACID is found on the following lists

Not Applicable

### Additional Regulatory Information

Not Applicable

### **National Inventory Status**

National Inventory	Status		
Australia - AIIC / Australia Non-	No (5-bromonicotinic acid)		
Industrial Use			
Canada - DSL	No (5-bromonicotinic acid)		
Canada - NDSL	No (5-bromonicotinic acid)		
China - IECSC	No (5-bromonicotinic acid)		
Europe - EINEC / ELINCS / NLP	Yes		
Japan - ENCS	No (5-bromonicotinic acid)		
Korea - KECI	No (5-bromonicotinic acid)		
New Zealand - NZIoC	Yes		
Philippines - PICCS	No (5-bromonicotinic acid)		
USA - TSCA	No (5-bromonicotinic acid)		
Taiwan - TCSI	Yes		
Mexico - INSQ	No (5-bromonicotinic acid)		
Vietnam - NCI	Yes		
Russia - FBEPH	Yes		
	Yes = All CAS declared ingredients are on the inventory		
Legend:	No = One or more of the CAS listed ingredients are not on the inventory. These		
	ingredients may be exempt or will require registration.		

### **SECTION 16 Other information**

Revision Date	20/06/2022
Initial Date	12/05/2005

### **SDS Version Summary:**

3D3 version Summary.			
Version	Date of Update	Sections Updated	
5.1	27/06/2017	Hazards identification - Classification, Ecological Information - Environmental, Handling and storage - Storage (storage requirement)	
6.1	20/06/2022	Expiration. Review and update	



#### 5-Bromonicotinic Acid

Revision Date: 20-06-22 Revision Number: 6.1

#### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

#### **Definitions and abbreviations**

PC - TWA: Permissible Concentration-Time Weighted Average

PC - STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit

IDLH: Immediately Dangerous to Life or Health Concentrations.

ES: Exposure Standard OSF: Odour Safety Factor

NOAEL: No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: Bio Concentration Factors BEI: Biological Exposure Index DNEL: Derived No-Effect Level

PNEC: Predicted no-effect concentration

MARPOL: International Convention for the Prevention of Pollution from Ships

IMSBC: International Maritime Solid Bulk Cargoes Code

IGC: International Gas Carrier Code IBC: International Bulk Chemical Code

AIIC: Australian Inventory of Industrial Chemicals

DSL: Domestic Substances List NDSL: Non-Domestic Substances List

IECSC: Inventory of Existing Chemical Substance in China

EINECS: European INventory of Existing Commercial chemical Substances

ELINCS: European List of Notified Chemical Substances

NLP: No-Longer Polymers

**ENCS: Existing and New Chemical Substances Inventory** 

KECI: Korea Existing Chemicals Inventory NZIoC: New Zealand Inventory of Chemicals

PICCS: Philippine Inventory of Chemicals and Chemical Substances

TSCA: Toxic Substances Control Act TCSI: Taiwan Chemical Substance Inventory INSQ: Inventario Nacional de Sustancias Químicas

NCI: National Chemical Inventory

FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

This Material Safety Data Sheet (MSDS) has been prepared in accordance with our company standards and is intended solely for the use of trained personnel. The information provided herein is believed to be accurate as of the date of issue, but no warranty, express or implied, is made regarding its accuracy, completeness, or suitability for any particular purpose. The user is responsible for ensuring safe conditions and compliance with applicable laws and regulations.