

2,3-Dichloro-5-trifluoromethyl-pyridine

Revision Date: 27-10-23 Revision Number: 6.1

### SECTION 1 Identification of the substance / mixture and of the company / undertaking

 Product Name
 2,3-DICHLORO-5-(TRIFLUOROMETHYL)PYRIDINE

 Chemical Name
 Not Available

**Synonyms** C6-H2-Cl2-F3-N; 2,3,5-DCTF

Proper shipping name ENVIRONMENTALLY HAZARDOUS SUBSTANCE, LIQUID, N.O.S. (contains 2,3-dichloro-5-

(trifluoromethyl)pyridine)

Chemical FormulaC6H2Cl2F3NOther means of identificationNot available

Other means of identification Not available CAS Number 69045-84-7

#### Relevant identified uses of the substance or mixture and uses advised against:

Relevant identified uses	Pyridines are heterocyclic six-membered aromatic compounds containing a single nitrogen atom. Pyridines are a class of important heterocycles and appear in many naturally occurring bioactive compounds, pharmaceutical molecules, and chiral ligands in polysubstituted forms. The pyridine moiety is present in countless molecules with applications as varied as catalysis, drug design, molecular recognition, and natural product synthesis. Pyridine derivatives have also been implicated as small molecule alpha-helical
	<ul> <li>mimetics in the inhibition of protein-protein interactions, and functionally selective GABAA ligands.</li> <li>Halogenated pyridines in particular are attractive building blocks for various cross-coupling methodologies, including Suzuki- Miyaura cross-coupling reactions</li> <li>Intermediate</li> </ul>

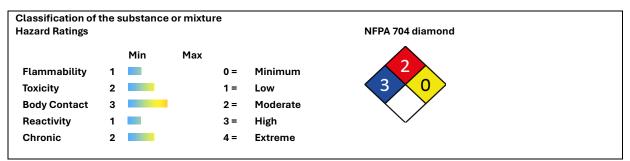
### $\label{lem:decomposition} \textbf{Details of the manufacturer or supplier of the safety data sheet:}$

<u> </u>	
Registered company name	Cohizon Life Sciences Limited
Address	Plot No. 6102/3, 6117-19, 5809-10, GIDC, Ankleshwar Gujarat 393002 India
Telephone	Not Available
Fax	Not Available
Website	Not Available
Email	Not Available

#### **Emergency telephone number:**

Association / Organisation	Cohizon Life Sciences Limited
Emergency telephone number(s)	(+91) 7046611150/51
Other emergency telephone number(s)	(+91) 6357684904

## **SECTION 2 Hazards identification**



Note: The hazard category numbers found in GHS classification in section 2 of this SDSs are NOT to be used to fill in the NFPA 704 diamond. Blue = Health Red = Fire Yellow = Reactivity White = Special (Oxidizer or water reactive substances).



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Classification	Flammable Liquids Category 4, Acute Toxicity (Oral and Inhalation) Category 4, Skin
	Corrosion/Irritation Category 3, Sensitisation (Skin) Category 1, Serious Eye
	Damage/Eye Irritation Category 1, Hazardous to the Aquatic Environment Long- Term
	Hazard Category 2

#### Label elements

Hazard pictogram(s)		
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	Signal word	Danger
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#### Hazard statement(s)

H227	Combustible liquid.	
H302+H332	Harmful if swallowed or if inhaled.	
H316	Causes mild skin irritation.	
H317	May cause an allergic skin reaction.	
H318	Causes serious eye damage.	
H411	Toxic to aquatic life with long lasting effects.	

#### Precautionary statement(s) Prevention

P210	Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No	
	smoking.	
P261	Avoid breathing mist/vapours/spray.	
P264	Wash all exposed external body areas thoroughly after handling.	
P270	Do not eat, drink or smoke when using this product.	
P271	Use only outdoors or in a well-ventilated area.	
P280	Wear protective gloves, protective clothing, eye protection and face protection.	
P273	Avoid release to the environment.	
P272	Contaminated work clothing should not be allowed out of the workplace.	

## Precautionary statement(s) Response

P301+P312	IF SWALLOWED: Call a POISON CENTER/doctor/physician/first aider if you feel unwell.	
P304+P340	IF INHALED: Remove person to fresh air and keep comfortable for breathing.	
P305+P351+P338	IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if	
	present and easy to do. Continue rinsing.	
P310	Immediately call a POISON CENTER/doctor/physician/first aider.	
P330	Rinse mouth.	
P370+P378	In case of fire: Use alcohol resistant foam or normal protein foam to extinguish.	
P302+P352	IF ON SKIN: Wash with plenty of water.	
P333+P313	If skin irritation or rash occurs: Get medical advice/attention.	
P362+P364	Take off contaminated clothing and wash it before reuse.	
P391	Collect spillage.	

#### Precautionary statement(s) Storage

F	P403	Store in a well-ventilated place.
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## Precautionary statement(s) Disposal

P501	Dispose of contents/container to authorised hazardous or special waste collection point	
	in accordance with any local regulation.	



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### SECTION 3 Composition / information on ingredients

### Substances

Cas No.	%[weight]	Name
69045-84-7	>98	2,3-dichloro-5-(trifluoromethyl)pyridine

#### **Mixtures**

See Section above for composition of substances

#### **SECTION 4 First aid measures**

Description of first aid measures

Description of first aid measures			
	If this product comes in contact with the eyes:		
Eye Contact	Wash out immediately with fresh running water.		
	<ul> <li>Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.</li> <li>Seek medical attention without delay; if pain persists or recurs seek medical</li> </ul>		
	<ul> <li>attention.</li> <li>Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.</li> </ul>		
	If skin or hair contact occurs:		
Skin Contact	<ul> <li>Immediately remove all contaminated clothing, including footwear.</li> <li>Flush skin and hair with running water (and soap if available</li> <li>Seek medical attention in event of irritation</li> </ul>		
	<ul> <li>If fumes or combustion products are inhaled remove from contaminated area.</li> <li>Lay patient down. Keep warm and rested.</li> </ul>		
Inhalation	<ul> <li>Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.</li> <li>Apply artificial respiration if not breathing, preferably with a demand valve</li> </ul>		
	resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.		
	Transport to hospital, or doctor.		
	IF SWALLOWED, REFER FOR MEDICAL ATTENTION, WHERE POSSIBLE, WITHOUT DELAY.		
	For advice, contact a Poisons Information Centre or a doctor.		
	Urgent hospital treatment is likely to be needed		
Ingestion	In the meantime, qualified first-aid personnel should treat the patient following observation and employing supportive measures as indicated by the patient's condition.		
	<ul> <li>If the services of a medical officer or medical doctor are readily available, the patient should be placed in his/her care and a copy of the SDS should be provided. Further action will be the responsibility of the medical specialist.</li> </ul>		
	If medical attention is not available on the worksite or surroundings send the patient to a hospital together with a copy of the SDS.		
	Where medical attention is not immediately available or where the		
	patient is more than 15 minutes from a hospital or unless instructed otherwise:		
	INDUCE vomiting with fingers down the back of the throat, ONLY IF		
	CONSCIOUS. Lean patient forward or place on left side (head-down		
	position, if possible) to maintain open airway and prevent aspiration.		
	NOTE: Wear a protective glove when inducing vomiting by mechanical means		
	1 0 0,		

#### Indication of any immediate medical attention and special treatment needed

As in all cases of suspected poisoning, follow the ABCDEs of emergency medicine (airway, breathing, circulation, disability, exposure), then the ABCDEs of toxicology (antidotes, basics, change absorption, change distribution, change elimination).

For poisons (where specific treatment regime is absent):



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#### **Basic Treatment:**

- Establish a patent airway with suction where necessary.
- Watch for signs of respiratory insufficiency and assist ventilation as necessary
- Administer oxygen by non-rebreather mask at 10 to 15 L/min
- Monitor and treat, where necessary, for pulmonary oedema
- Monitor and treat, where necessary, for shock
- · Anticipate seizures
- DO NOT use emetics. Where ingestion is suspected rinse mouth and give up to 200 ml water (5 ml/kg recommended) for dilution where patient is able to swallow, has a strong gag reflex and does not drool.

#### **Advanced Treatment:**

- Consider orotracheal or nasotracheal intubation for airway control in unconscious patient or where respiratory arrest has occurred.
   Positivepressure ventilation using a bag-valve mask might be of use.
- Monitor and treat, where necessary, for arrhythmias.
- Start an IV D5W TKO. If signs of hypovolaemia are present use lactated Ringers solution. Fluid overload might create complications. Drug therapy should be considered for pulmonary oedema.
- Hypotension with signs of hypovolaemia requires the cautious administration of fluids. Fluid overload might create complications.
   Treat seizures with diazepam.
- Proparacaine hydrochloride should be used to assist eye irrigation.

BRONSTEIN, A.C. and CURRANCE, P.L.

EMERGENCY CARE FOR HAZARDOUS MATERIALS EXPOSURE: 2nd Ed. 1994

Treat symptomatically

## **SECTION 5 Firefighting measures**

## Extinguishing media

- Foam
- Dry Chemical Powder
- BCF (where regulations permit).
- Carbon dioxide
- Water Spray or Fog Larges files only

## Special hazards arising from the substrate or mixture

Fire Incompatibility	Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches,
	pool chlorine etc. as ignition may result.

#### Advice for firefighters

Advice for in engineers		
	<ul> <li>Alert Fire Brigade and tell them location and nature of hazard.</li> </ul>	
	<ul> <li>Wear breathing apparatus plus protective gloves</li> </ul>	
	<ul> <li>Prevent, by any means available, spillage from entering drains or water courses</li> </ul>	
	Fight fire from a safe distance, with	
Fire Fighting	adequate cover.	
	Use water delivered as a fine spray to control fire and cool adjacent area	
	DO NOT approach containers suspected to be hot	
	Cool fire exposed containers with water spray from a protected location	
	<ul> <li>If safe to do so, remove containers from path of fire.</li> </ul>	
	Combustible.	
	Slight fire hazard when exposed to heat or flame.	
Fire/Evaluation Horard	Heating may cause expansion or	
Fire/Explosion Hazard	decomposition leading to violent rupture of	
	containers.	
	On combustion, may emit toxic fumes of	



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	carbon monoxide (CO).
•	May emit acrid smoke.
•	Mists containing combustible materials may be explosive.
Combu	stion products include:
•	carbon dioxide (CO2)
•	Hydrogen Chloride
•	Phosgene
•	Hydrogen Fluoride
•	nitrogen oxides (NOx)
•	other pyrolysis products typical of burning organic material.

## SECTION 6 Accidental release measures

Personal precautions, protective equipment and emergency procedures

See section 8

**Environmental precautions** 

See section 12

Methods and material for containment and cleaning up

Methods and material for containment and cleaning up		
Minor Spills	Remove all ignition sources.	
	Clean up all spills immediately.	
	Avoid contact with skin and eyes.	
	Control personal contact with the substance, by using protective equipment.	
	Use dry clean up procedures and avoid generating dust.	
	Place in a suitable, labelled container for waste disposal	
	Moderate hazard.	
	CAUTION: Advise personnel in area.	
	<ul> <li>Alert Emergency Services and tell them the location and nature of hazard.</li> </ul>	
	<ul> <li>Control personal contact by wearing protective clothing.</li> </ul>	
	<ul> <li>Prevent, by any means available, spillage from entering drains or water courses</li> </ul>	
Major Spills	Recover products wherever possible.	
Major Spills	IF DRY: Use dry clean up procedures and avoid generating dust. Collect	
	residues and place in sealed plastic bags or other containers for disposal.	
	<ul> <li>IF WET: Vacuum/shovel up and place in labelled containers for disposal.</li> </ul>	
	ALWAYS: Wash area down with large amounts of water and prevent runoff into	
	drains.	
	<ul> <li>If contamination of drains or waterways occurs, advise Emergency Services</li> </ul>	

Personal Protective Equipment advice is contained in Section 8 of the SDS

## SECTION 7 Handling and storage

Precautions for	or safe handling
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Frecautions for safe manutin	5
	<ul> <li>Avoid all personal contact, including inhalation.</li> </ul>
	Wear protective clothing when risk of exposure occurs.
	Use in a well-
	ventilated area.
	Prevent concentration in hollows and sumps.
	DO NOT enter confined spaces until the atmosphere has been checked.
Safe Handling	DO NOT allow material to contact humans, exposed food or food utensils
	Avoid contact with incompatible
	materials.
	When handling, <b>DO NOT</b> eat, drink or smoke.
	Keep containers securely sealed when not in use.
	Avoid physical damage to containers
	Always wash hands with soap and water after handling



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	Work clothes should be laundered separately
	Use good occupational work practice
	<ul> <li>Observe manufacturer's storage and handling recommendations contained within this SDS.</li> </ul>
	<ul> <li>Atmosphere should be regularly checked against established exposure</li> </ul>
	standards to ensure safe working conditions.
	Store in original containers.
	Keep containers securely sealed.
	Store in a cool, dry area protected from environmental extremes.
Other information	<ul> <li>Store away from incompatible materials and foodstuff containers.</li> </ul>
	<ul> <li>Protect containers against physical damage and check regularly for leaks.</li> </ul>
	Observe manufacturer's storage and handling recommendations contained within this SDS

#### Conditions for safe storage, including any incompatibilities

Conditions for safe storage, in	cluding any incompatibilities	
	Metal can or drum	
Suitable container	Packaging as recommended by manufacturer.	
	<ul> <li>Check all containers are clearly labelled and free from leaks</li> </ul>	
	Contains a six-membered heterocyclic ring.	
	Six-membered heterocycles can be described as pideficient. Substitution	
	by electronegative groups or additional nitrogen atoms in the ring	
	significantly increase the pi-deficiency. These effects also decrease the	
	basicity.	
	<ul> <li>Electrophilic aromatic substitution is more difficult</li> </ul>	
	while nucleophilic aromatic substitution is facilitated.	
	for pyridines:	
	Because of the electronegative nitrogen in the pyridine ring, the molecule is	
	relatively electron deficient. It, therefore, enters less readily electrophilic	
	aromatic substitution reactions, which are characteristic of benzene	
	derivatives; even more so if the reaction mix doesn't scavenge protons	
	released by the reaction (protonated pyridine is even more electron-	
	deficient). However, unlike benzene and its derivatives, pyridine is more	
	prone to nucleophilic substitution and metalation of the ring by strong	
	organometallic bases.	
	<ul> <li>The nitrogen center of pyridine features a basic lone pair of electrons.</li> </ul>	
Storage incompatibility	Because this lone pair is not part of the aromatic ring, pyridine is a base,	
otorago moompatizatio,	having chemical properties similar to those of tertiary amines. Pyridine can	
	act as Lewis base, donating its pair of electrons to a Lewis acid.	
	<ul> <li>Pyridine is protonated by reaction with acids and forms a positively</li> </ul>	
	charged aromatic polyatomic ion called pyridinium	
	<ul> <li>The reactivity of pyridine can be distinguished for three chemical groups.</li> </ul>	
	<ul> <li>With electrophiles, electrophilic substitution takes place where pyridine</li> </ul>	
	expresses aromatic properties.	
	<ul> <li>With nucleophiles, pyridine reacts at positions 2 and 4 and thus behaves</li> </ul>	
	similar to imines and carbonyls.	
	The reaction with many Lewis acids results in the addition to the nitrogen atom	
	of pyridine, which is similar to the reactivity of tertiary amines. The ability of	
	pyridine and its derivatives to oxidize, forming amine oxides (N-oxides), is also	
	a feature of tertiary amines	
	<ul> <li>Secondary amines form salts with strong acids and can be oxidized to the</li> </ul>	
	corresponding nitrone using hydrogen peroxide, catalyzed by selenium dioxide.	
	<ul> <li>Avoid strong acids, bases.</li> </ul>	
	Avoid reaction with oxidising agents	



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## SECTION 8 Exposure controls / personal protection

#### **Control Parameters**

Occupational Exposure Limits (OEL)

### INGREDIENT DATA

Not Available

Ingredient	Original IDLH	Revised IDLH
2,3-dichloro-5-	Not Available	Not Available
(trifluoromethyl)pyridine		

## Occupational Exposure Banding

Ingredient	Occupational Exposure Band Rating	Occupational Exposure Band Limit
2,3-dichloro-5-	E	<0.1 ppm
(trifluoromethyl)pyridine		
Notes:	Occupational exposure banding is a process of assigning chemicals into specific categories or bands based on a chemical's potency and the adverse health outcomes associated with exposure. The output of this process is an occupational exposure band (OEB), which corresponds to a range of exposure concentrations that are expected to protect worker health.	

0	f exposure concentrations that are expected to protect worker health.	
Exposure controls		
Appropriate engineering controls	<ul> <li>Engineering controls are used to remove a hazard or between the worker and the hazard. Well-designed econtrols can be highly effective in protecting workers be independent of worker interactions to provide this protection.</li> <li>The basic types of engineering controls are:</li> <li>Process controls which involve changing the way a joing is done to reduce the risk.</li> <li>Enclosure and/or isolation of emission source which hazard "physically" away from the worker and ventilates strategically "adds" and "removes" air in the work enviolent to the ventilation can remove or dilute an air contaminant in properly. The design of a ventilation system must may process and chemical or contaminant in use. Emplow use multiple types of controls to prevent employee or buse to be calculated an expression.</li> <li>Local exhaust ventilation usually required. If risk of onexists, wear approved respirator. Correct fit is essential adequate protection. Supplied-air type respirator may in special circumstances. Correct fit is essential adequate protection.</li> <li>An approved self contained breathing apparatus (SCI some situations.</li> <li>Provide adequate ventilation in warehouse or closcontaminants generated in the workplace posse velocities which, in turn, determine the "capture circulating air required to effectively remove the contaminant of the contaminant in the contaminant in the capture circulating air required to effectively remove the contaminant in turn.</li> </ul>	engineering and will typically s high level of  b activity or process  keeps a selected tion that vironment. If designed toth the particular yers may need to verexposure ial to obtain y be required I to ensure  BA) may be required in sed storage area. Air ses varying "escape' e velocities" of fresh taminant
	Type of Contaminant	Air Speed
	solvent, vapours, degreasing etc., evaporating from tank (in still air).	0.25-0.5 m/s (50- 100 f/min.)
	aerosols, fumes from pouring operations, intermittent container filling, low speed conveyer transfers, welding, spray drift, plating acid fumes, pickling (released at low velocity into	0.5-1 m/s (100- 200 f/min.)

zone of active generation)

generation into zone of rapid air motion)

direct spray, spray painting in shallow booths, drum filling,

conveyer loading, crusher dusts, gas discharge (active

1-2.5 m/s (200-

500 f/min.)



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	grinding, abrasive blasting, tumbling, high	n speed wheel	2.5-10 m/s (500-		
	generated dusts (released at high initial velocity into zone of 2000 f/min.)				
	very high rapid air motion).				
	Which each range the appropriate value depends on				
	Lower end of the range	Upper end of the	range		
	Room air currents minimal or favourable to capture	1. Disturbing roo	om air currents		
	Contaminants of low toxicity or of nuisance value only	2: Contaminants	s of high toxicity		
	3. Intermittent, low production	3: High production, heavy use			
	4. Large hood or large air mass in motion	4: Small hood-lo	cal control only		
	<ul> <li>Simple theory shows that air velocit</li> </ul>	y falls rapidly with	distance away from		
	the opening of a simple extraction p	ipe. Velocity gene	rally decreases with		
	the square of distance from the extr	action point (in sin	nple cases).		
	Therefore, the air speed at the extra	ction point should	l be adjusted,		
	accordingly, after reference to dista	nce from the conta	aminating source.		
	The air velocity at the extraction far		•		
	of 4-10 m/s (800-2000 ft/min) for ex	•			
	metres distant from the extraction p		=		
	considerations, producing performa				
	apparatus, make it essential that the				
Individual protection	factors of 10 or more when extraction	ni systems are ms	statted of used.		
•					
measures, such as personal					
protective		37			
equipment	000				
	<ul> <li>Safety glasses with side shields.</li> <li>Chemical goggles. [AS/NZS 1337.1,</li> <li>Contact lenses may pose a special</li> </ul>				
	absorb and concentrate irritants. A written policy document, describing				
	the wearing of lenses or restrictions on use, should be created for each				
	workplace or task. This should include a review of lens absorption and				
	adsorption for the class of chemicals in use and an account of injury				
	experience.				
Eye and face protection	·				
	Medical and first-aid personnel should be trained in their removal a				
	suitable equipment should be readi	ly available. In the	event of		
	chemical exposure, begin eye irriga	tion immediately a	and remove		
	contact lens as soon as practicable	. Lens should be r	removed at the		
	first signs of eye redness or irritation	n - lens should be r	removed in a		
	clean environment only after worke				
	-		mao morouginy.		
Oldin manda aki am	[CDC NIOSH Current Intelligence Bulletin 59].				
Skin protection	See Hand protection below				
	Wear chemical protective gloves, e.g. PVC.				
	Wear safety footwear or safety gumboots, e.g	g. Rubber			
	NOTE:				
	The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other				
	protective equipment, to avoid all p	ossible skin conta	ct.		
Hands/feet protection	<ul> <li>Contaminated leather items, such a should be removed and destroyed</li> </ul>	as shoes, belts an	d watch-bands		
	The selection of suitable gloves does not only	depend on the ma	aterial, but also		
	on further marks of quality which vary from m	anufacturer to ma	nufacturer.		
	Where the chemical is a preparation of severa	al substances, the	resistance of		
	the glove material can not be calculated in ad	lvance and has the	erefore to be		
	checked prior to the application.				
	The exact break through time for substances	has to be obtained	from the		
	1 3 11 11 11 11 11 11 11 11 11 11 11 11		-		



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manufacturer of the protective gloves and has to be observed when making a final choice.

Personal hygiene is a key element of effective hand care. Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturizer is recommended.

Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include:

- frequency and duration of contact,
- chemical resistance of glove material,
- glove thickness and
- dexterity

Select gloves tested to a relevant standard (e.g. Europe EN 374, US F739, AS/NZS 2161.1 or national equivalent).

- When prolonged or frequently repeated contact may occur, a glove with a protection class of 5 or higher (breakthrough time greater than 240 minutes according to EN 374, AS/NZS 2161.10.1 or national equivalent) is recommended.
- When only brief contact is expected, a glove with a protection class of 3 or higher (breakthrough time greater than 60 minutes according to EN 374, AS/NZS 2161.10.1 or national equivalent) is recommended.
- Some glove polymer types are less affected by movement and this should be taken into account when considering gloves for long-term
- Contaminated gloves should be replaced.

As defined in ASTM F-739-96 in any application, gloves are rated as:

- Excellent when breakthrough time > 480 min
- Good when breakthrough time > 20 min
- Fair when breakthrough time < 20 min</li>
- Poor when glove material degrades

For general applications, gloves with thickness typically greater than 0.35 mm, are recommended.

It should be emphasized that glove thickness is not necessarily a good predictor of glove resistance to a specific chemical, as the permeation efficiency of the glove will be dependent on the exact composition of the glove material. Therefore, glove selection should also be based on consideration of the task requirements and knowledge of breakthrough times.

Glove thickness may also vary depending on the glove manufacturer, the glove type and the glove model. Therefore, the manufacturers' technical data should always be taken into account to ensure selection of the most appropriate glove for the task. Note: Depending on the activity being conducted, gloves of varying thickness may be required for specific tasks. For example:

- Thinner gloves (down to 0.1 mm or less) may be required where a high
  degree of manual dexterity is needed. However, these gloves are only
  likely to give short duration protection and would normally be just for
  single use applications, then disposed of.
- Thicker gloves (up to 3 mm or more) may be required where there is a mechanical (as well as a chemical) risk i.e. where there is abrasion or puncture potential

Gloves must only be worn on clean hands. After using gloves, hands should be washed and dried thoroughly. Application of a non-perfumed moisturiser is recommended.

Neoprene gloves

Body protection	See Other protection below
Other protection	Overalls.



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P.V.C apron
Barrier cream.
Skin cleansing cream.
Eye wash unit

#### **Respiratory protection**

Type A Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent). Selection of the Class and Type of respirator will depend upon the level of breathing zone contaminant and the chemical nature of the contaminant. Protection Factors (defined as the ratio of contaminant outside and inside the mask) may also be important.

Required minimum protection factor	Maximum gas/vapour concentration present in air p.p.m. (by volume)	Half-face Respirator	Full-Face Respirator
up to 10	1000	A-AUS / Class1	-
up to 50	1000	-	A-AUS / Class 1
up to 50	5000	Airline *	-
up to 100	5000	-	A-2
up to 100	10000	-	A-3
100+			Airline**

<sup>\*\*</sup>Continuous Flow \*\* - Continuous-flow or positive pressure demand

A(All classes) = Organic vapours, B AUS or B1 = Acid gasses, B2 = Acid gas or hydrogen cyanide(HCN), B3 = Acid gas or hydrogen cyanide(HCN), E = Sulfur dioxide(SO2), G = Agricultural chemicals, K = Ammonia(NH3), Hg = Mercury, NO = Oxides of nitrogen, MB = Methyl bromide, AX = Low boiling point organic compounds(below 65 degC)

- Cartridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content.
- The wearer must be warned to leave the contaminated area immediately on detecting any odours
  through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour
  concentration is too high, or that the mask is not properly fitted. Because of these limitations, only
  restricted use of cartridge respirators is considered appropriate.
- Cartridge performance is affected by humidity. Cartridges should be changed after 2 hr of continuous use
  unless it is determined that the humidity is less than 75%, in which case, cartridges can be used for 4 hr. Used
  cartridges should be discarded daily, regardless of the length of time used

### **SECTION 9 Physical and chemical properties**

Information on basic physical and chemical properties

Appearance	·	n pungent pyridine-like odour; de ensity 0.015 g/l. Hydrolyses at le	oes not mix well with water (380 ow pH to form water soluble
Physical state	Liquid	Relative density (Water = 1)	1.549
Odour	Not Available	Partition coefficient n-	3.45
Odour threshold	Not Available	octanol / water Auto-ignition temperature	Not Available
pH (as supplied)	Not Applicable	(°C) Decomposition temperature (°C)	>300
Melting point / freezing point (°C)	8-9℃	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	169°C	Molecular weight (g/mol)	215.99
Flash point (°C)	79.44℃	Taste	Not Available
Evaporation rate	Not Available	<b>Explosive properties</b>	Not Available
Flammability	Combustible	Oxidising properties	Not Available



2,3-Dichloro-5-trifluoromethyl-pyridine

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Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	1.13 mm Hg (25 C)	Gas group	Not Available
Solubility in water	Partially miscible	pH as a solution (1%)	Not Applicable
Vapour density (Air = 1)	>1	VOC g/L	1519.57
Heat of Combustion (kJ/g)	Not Available	Ignition Distance (cm)	Not Available
Flame Height (cm)	Not Available	Flame Duration (s)	Not Available
<b>Enclosed Space Ignition Time</b>		Enclosed Space	
Equivalent (s/m3)	Not Available	Ignition Deflagration	Not Available
		Density (g/m3)	

## **SECTION 10 Stability and reactivity**

Reactivity	See section 7
Chemical stability	Unstable in the presence of incompatible materials
	Product is considered stable
	Hazardous polymerisation will not occur
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

## SECTION 11 Toxicological information

## Information on toxicological effects

	There is some evidence to suggest that the material can cause
	respiratory irritation in some persons. The body's response to such
Inhaled	irritation can cause further lung damage.
Innated	<ul> <li>Inhalation hazard is increased at higher temperatures.</li> </ul>
	<ul> <li>Inhalation of aerosols (mists, fumes), generated by the material during the</li> </ul>
	course of normal handling, may be harmful
	Accidental ingestion of the material may be harmful; animal experiments
Ingestion	indicate that ingestion of less than 150 gram may be fatal or may produce
	serious damage to the health of the individual.
	There is some evidence to suggest that this material can cause
	inflammation of the skin on contact in some persons. Open cuts, abraded
	or irritated skin should not be exposed to this material.
Skin Contact	<ul> <li>Entry into the blood-stream, through, for example, cuts, abrasions or</li> </ul>
	lesions, may produce systemic injury with harmful effects.
	Examine the skin prior to the use of the material and ensure that any
	external damage is suitably protected.
Eye	If applied to the eyes, this material causes severe eye damage.
	Skin contact with the material is more likely to cause a sensitization
	reaction in some persons compared to the general population.
	<ul> <li>There has been some concern that this material can cause cancer or</li> </ul>
	mutation but there is not enough data to make an assessment.
	Substance accumulation, in the human body, may occur and may
Chronic	cause some concern following repeated or long-term occupational
	exposure.
	Data from experimental studies indicate that pyridines represent a
	potential cause of cancer in man. They have also been shown
	<ul> <li>to cross the placental barrier in rats and cause premature delivery,</li> </ul>
	miscarriages and stillbirths.



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2,3-Dichloro-5-	TOXICITY	IRRITATION			
trifluoromethyl-pyridine	Dermal (rabbit) LD50:	Not Available			
	>1000<=2000 mg/kg <sup>[1]</sup>				
	Oral (Rat) LD50: >500<1000				
	mg/kg <sup>[1]</sup>				
Legend:		ppe ECHA Registered Substances - Acute toxicity			
		nufacturer's SDS. Unless otherwise specified data			
	extracted from RTECS - Reg	gister of Toxic Effect of chemical Substances			
	The following information refers to contact allergens as a group and may not b specific to this product.				
	Contact allergies quickly manifest themselves as contact eczema, more rarely as				
	urticaria or Quincke's oedema. The pathogenesis of contact eczema involves a				
	cell-mediated (T lymphocyt	es) immune reaction of the delayed type. Other			
	allergic skin reactions, e.g. contact urticaria, involve antibody-mediated immun				
2,3-Dichloro-5-	reactions. The significance of the contact allergen is not simply determined by its				
trifluoromethyl-pyridine	sensitisation potential: the	sensitisation potential: the distribution of the substance and the opportunities for			
	contact with it are equally important. A weakly sensitising substance which is				
	more important allergen than one with stronger				
	sensitising potential with which few individuals come into contact. From a clinical				
	are noteworthy if they produce an allergic test reaction				
	in more than 1% of the pers	sons tested.			
	No significant acute toxico	logical data identified in literature search.			

Acute Toxicity	~	Carcinogenicity	×	
Skin Irritation/Corrosion	~	Reproductivity	×	
Serious Eye Damage/Irritation	•	STOT - Single Exposure	×	
Respiratory or Skin sensitisation	•	STOT - Repeated Exposure	×	
Mutagenicity	×	Aspiration Hazard	×	
<b>Legand: X</b> − Data either not available or does not fill the criteria for classification				
√– Data available to make classification				

### **SECTION 12 Ecological information**

#### Toxicity

TOXICITY		1	_	1	
2,3-Dichloro-5-trifluoromethyl-	End Point	Test Duration (hr.)	Species	Value	Source
	EC50	72h	Algae or other aquatic plants	2.9mg/l	2
pyridine	NOEC(ECx)	72h	Algae or other aquatic plants	0.28mg/l	2
	LC50	96h	Fish	3.9mg/l	2
	EC50	48h	Crustacea	4.9mg/l	2
Legand:	Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 8. Vendor Data				

Bioconcentration potential is moderate (BCF between 100-3000 or Log Pow between 3-5). Log octanol/water partition coefficient (Log Pow) is estimated to be 3.45. Potential for mobility in soil is low (Koc between 500-2000). Log soil organic carbon partition coefficient (Log Koc) is estimated to be 2.87. Log air/water partition coefficient (Log Kaw) is - 1.33. Degradation and Persistance Biodegradation under aerobic laboratory conditions is below detectable limits



#### 2,3-Dichloro-5-trifluoromethyl-pyridine

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(<2.5%). Theoretical oxygen demand (ThOD) is calculated to be 0.954. Ecotoxicity: Material is moderately toxic to aquatic organisms on an acute basis (LC50 or EC50 is between 1-10 mg/L in most sensitive species). Fish LC50: fathead minnow (Pimephales promelas) 13.4 mg/l, rainbow trout (Oncorhyncus mykiss) 3.9 mg/l Dahpnia magna LC50: 31.9 mg/l

Toxic to aquatic organisms, may cause long-term adverse effects in the aquatic environment.

Do NOT allow product to come in contact with surface waters or to intertidal areas below the mean high water mark. Do not contaminate water when cleaning equipment or disposing of equipment wash-waters.

Wastes resulting from use of the product must be disposed

of on site or at approved waste sites. For Pyridine and its

#### Derivatives:

Environmental Fate: As molecular weight/substitution increase in the pyridine category, greater distribution to water and soil, and less to air, is predicted. Atmospheric Fate: The lower weight pyridine, piperidine, is expected to be rapidly degraded by UV light in the atmosphere, with an estimated half-life of < 1 day. Higher molecular weight pyridines are expected to be broken down by sunlight, (photodegrades), more slowly, (half-lives ranging from 10-30 days). Lutidines and collidines are expected to photodegrade even more slowly. The nitrile derivatives of pyridine are also predicted to photodegrade slowly, with half-lives of 164 days; however, the nitrile derivatives of pyridine are predicted to partition to air much less favorably than to soil and water.

Terrestrial Fate: Depending upon the environmental conditions, different types of bacteria, fungi, and enzymes are involved in the breakdown of these substances. Aquatic Fate: Pyridines are not expected to be broken down by water; however, breakdown by microbes is expected, if sufficient oxygen is available. These substances are expected to be stable in low oxygen/sterile conditions.

Ecotoxicity: Pyridine and its derivatives range from slightly to moderately toxic to fish, invertebrates and algae.

#### DO NOT discharge into sewer or waterways

Ingredient	Persistence: Water/Soil	Persistence: Air
2,3-Dichloro-5-trifluoromethyl- pyridine	High	High

#### Bio accumulative potential

bio accumutative potentiat	
Ingredient	Bioaccumulation
2,3-Dichloro-5-trifluoromethyl- pyridine	Low (BCF = 3.06)

#### Mobility in soil

Mobility in Soil	
Ingredient	Mobility
2,3-Dichloro-5-trifluoromethyl-	LOW (Log KOC = 3162)

## **SECTION 13 Disposal considerations**

#### Waste treatment methods

Product / Packaging disposal	<ul> <li>Containers may still present a chemical hazard/ danger when empty</li> <li>Return to supplier for reuse/ recycling if possible.</li> <li>Otherwise:</li> <li>If container cannot be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill</li> <li>Where possible retain label warnings and SDS and observe all notices pertaining to the product</li> <li>Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area. In some areas, certain wastes must be tracked.</li> <li>A Hierarchy of Controls seems to be common - the user should investigate:</li> </ul>
	A Hierarchy of Controls seems to be common - the user should investigate:
	Reduction
	• Reuse



2,3-Dichloro-5-trifluoromethyl-pyridine

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- Recycling
- Disposal (if all else fails)
- This material may be recycled if unused, or if it has not been contaminated so as to make it unsuitable for its intended use. If it has been contaminated, it may be possible to reclaim the product by filtration, distillation or some other means. Shelf life considerations should also be applied in making decisions of this type. Note that properties of a material may change in use, and recycling or reuse may not always be appropriate.
- DO NOT allow wash water from cleaning or process equipment to enter
- It may be necessary to collect all wash water for treatment before disposal.
- In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.
- Where in doubt contact the responsible authority.
- Recycle wherever possible or consult manufacturer for recycling options.
- Consult State Land Waste Authority for disposal.
- Bury or incinerate residue at an approved site.
- Recycle containers if possible, or dispose of in an authorised landfill

#### **SECTION 14 Transport information**

Labels Required	
Marine Pollutant	¥2

#### Land transport (UN):

14.1. UN number or ID number	3082	
14.2. UN proper shipping	ENVIRONMENTALLY HAZARDOUS SUBSTANCE, LIQUID, N.O.S. (contains 2,3-	
name	dichloro-5-(trifluoromethyl)pyridine)	
14.3. Transport hazard	Class	9
class(es)	Subsidiary Hazard	Not Applicable
14.4. Packing group	III	
14.5. Environmental hazard	Environmentally hazardous	
14.6. Special precautions for	Special provisions	274; 331; 335; 375
user	Limited quantity	5 L

#### Air transport (ICAO-IATA / DGR):

14.1. UN number	3082	
14.2. UN proper shipping	ENVIRONMENTALLY HAZARDOUS SUBSTANCE, LIQUID, N.O.S. (contains 2,3-	
name	dichloro-5-(trifluoromethyl)pyridine)	
14.3. Transport hazard	ICAO/IATA Class	9
class(es)	ICAO / IATA Subsidiary Hazard	Not Applicable
	ERG Code	9L
14.4. Packing group	III	
14.5. Environmental hazard	Environmentally hazardous	
14.6. Special precautions for	Special provisions	A97 A158 A197 A215
user	Cargo Only Packing Instructions	964
	Cargo Only Maximum Qty / Pack	450 L
	Passenger and Cargo Packing Instructions	964
	Passenger and Cargo Maximum Qty / Pack	450 L



## 2,3-Dichloro-5-trifluoromethyl-pyridine

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Passenger and Cargo Limited Quantity Packing Instructions	Y964
Passenger and Cargo Limited Maximum Qty / Pack	30 kg G

#### Sea transport (IMDG-Code/ GGVSee):

14.1. UN number	3082	
14.2. UN proper shipping	ENVIRONMENTALLY HAZARDOUS SUBSTANCE, LIQUID, N.O.S. (contains 2,3-	
name	dichloro-5-(trifluoromethyl)pyridine)	
14.3. Transport hazard	IMDG Class	9
class(es)	IMDG Subsidiary Hazard	Not Applicable
14.4. Packing group	III	
14.5. Environmental hazard	Marine Pollutant	
14.6. Special precautions for	EMS Number	F-A,S-F
user	Special provisions	274 335 969
	Limited Quantities	5 L

#### 14.7.1 Transport in bulk in accordance to Annex II of MARPOL and the IBC Code

Not Applicable

#### 14.7.2 Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product Name	Ship Type
2,3-dichloro-5-	Not Available
(trifluoromethyl)pyridine	

#### 14.7.3 Transport in bulk in accordance with the IGC Code

14:7:0 Transport in bank in accordance with the 100 Code	
Product Name	Ship Type
2,3-dichloro-5-	Not Available
(trifluoromethyl)pyridine	

## SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

### 2,3-dichloro-5-(trifluoromethyl)pyridine is found on the following regulatory lists

Not Applicable

#### Additional Regulatory Information

Not Applicable

#### **National Inventory Status**

National Inventory	Status
Australia - AIIC / Australia Non-	No (2,3-dichloro-5-(trifluoromethyl)pyridine)
Industrial Use	
Canada - DSL	No (2,3-dichloro-5-(trifluoromethyl)pyridine)
Canada - NDSL	No (2,3-dichloro-5-(trifluoromethyl)pyridine)
China - IECSC	Yes
Europe - EINEC / ELINCS / NLP	Yes
Japan - ENCS	Yes
Korea - KECI	Yes
New Zealand - NZIoC	No (2,3-dichloro-5-(trifluoromethyl)pyridine)
Philippines - PICCS	No (2,3-dichloro-5-(trifluoromethyl)pyridine)
USA - TSCA	All chemical substances in this product have been designated as TSCA Inventory 'Active'
Taiwan - TCSI	Yes
Mexico - INSQ	No (2,3-dichloro-5-(trifluoromethyl)pyridine)
Vietnam - NCI	Yes
Russia - FBEPH	No (2,3-dichloro-5-(trifluoromethyl)pyridine)
	Yes = All CAS declared ingredients are on the inventory
Legend:	No = One or more of the CAS listed ingredients are not on the inventory. These
	ingredients may be exempt or will require registration.

## **SECTION 16 Other information**

Revision Date	27/10/2023
Initial Date	12/05/2005



2,3-Dichloro-5-trifluoromethyl-pyridine

Revision Date: 27-10-23 Revision Number: 6.1

#### **SDS Version Summary:**

Version	Date of Update	Sections Updated
		Handling and storage - Storage (storage incompatibility),
5.1	15/09/2019	Identification of the substance / mixture and of the company
		/ undertaking - Use
6.1	27/10/2023	UN Number update

#### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

#### **Definitions and abbreviations**

PC - TWA: Permissible Concentration-Time Weighted Average

PC - STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit

IDLH: Immediately Dangerous to Life or Health Concentrations.

ES: Exposure Standard OSF: Odour Safety Factor

NOAEL: No Observed Adverse Effect Level LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value LOD: Limit Of Detection OTV: Odour Threshold Value BCF: Bio Concentration Factors BEI: Biological Exposure Index DNEL: Derived No-Effect Level

PNEC: Predicted no-effect concentration

MARPOL: International Convention for the Prevention of Pollution from Ships

IMSBC: International Maritime Solid Bulk Cargoes Code

IGC: International Gas Carrier Code IBC: International Bulk Chemical Code

AIIC: Australian Inventory of Industrial Chemicals

DSL: Domestic Substances List

NDSL: Non-Domestic Substances List

IECSC: Inventory of Existing Chemical Substance in China

EINECS: European INventory of Existing Commercial chemical Substances

ELINCS: European List of Notified Chemical Substances

NLP: No-Longer Polymers

ENCS: Existing and New Chemical Substances Inventory

KECI: Korea Existing Chemicals Inventory NZIoC: New Zealand Inventory of Chemicals

PICCS: Philippine Inventory of Chemicals and Chemical Substances

TSCA: Toxic Substances Control Act

TCSI: Taiwan Chemical Substance Inventory INSQ: Inventario Nacional de Sustancias Químicas

NCI: National Chemical Inventory

FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

This Material Safety Data Sheet (MSDS) has been prepared in accordance with our company standards and is intended solely for the use of trained personnel. The information provided herein is believed to be accurate as of the date of issue, but no warranty, express or implied, is made regarding its accuracy, completeness, or suitability for any particular purpose. The user is responsible for ensuring safe conditions and compliance with applicable laws and regulations.